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Exam 3

- Time:
 - Tuesday, November 8: 7:00 9:00PM OR
 - Wednesday, November 9: 7:00 9:00PM OR
 - Thursday, November 10: 7:00 10:00PM
- Location Soc/Anthro Testing Center
 - Chapters will be covered in this order: Chapter 17, 18
- Practice Exams are Posted
 - Ex3A Practice Exam 3A
 - Ex3B Practice Exam 3B
- Deadline for alternate arrangements is Monday, 11/7/2016 at 4:30 PM (i.e., close of business)
 - An oral make-up exam will be required for making up the exam for all students not taking the exam on the above dates or having already made prior arrangements

| Ex3-01-B7-17-01A Ketone Aldehyde Naming | | Friday, October 22 | |
|---|----------------------------------|-----------------------|--|
| Ex3-01-B7-17-01B Aldehyde Ketone Naming | | Saturday, October 22 | |
| Ex3-02-B7-17-02A Ald Ket Rxns O-Nucl | | Saturday, October 23 | |
| Ex3-02-B7-17-02B Ald Ket O-Nucleophiles | | Sunday, October 24 | |
| Ex3-02-B7-17-02C Ald Ket Rxns O-Nucl | | Monday, October 25 | |
| Ex3-03-B7-17-03A Ald Ket with N-Nucl | | Tuesday, October 26 | |
| | 3-B7-17-03B Ald Ket with N-Nucl | Wednesday, October 27 | |
| Exam 3 Ex3-0 | 3-B7-17-03C Ald Ket with N-Nucl | Thursday, October 28 | |
| Lecture Ex3-0 | 4-B7-17-04A Ald Ket with C-Nucl | Friday, October 29 | |
| Fv3-01-R7-17-01R Ald Ket with C-Nucl | | Saturday, October 29 | |
| Planning Ex3-04-B7-17-04C Ald Ket with C-Nucl | | Sunday, October 30 | |
| Ex3-05-B7-18-01 Tautomers | | Sunday, October 30 | |
| Ex3-0 | 6-B7-18-02B Alpha-Bromination | Monday, October 31 | |
| Ex3-0 | 6-B7-18-02C Alpha-Bromination | Tuesday, November 1 | |
| Ex3-07 | 7-B7-18-03B Alkylation Alpha-C=O | Wednesday, November 2 | |
| Ex3-07 | 7-B7-18-03C Alkylation Alpha-C=O | Thursday, November 3 | |
| Ex3-00-D7-10-04D Walonic Ester Synthesis | | rnuay, November 4 | |
| Ex3-08-B7-18-04C Malonic Ester Synthesis | | Saturday, November 5 | |
| Ex3-09-B7-18-05 Fatty Acids | | Sunday, November 6 | |
| Exam 3 | | November 8, 9, 10 | |

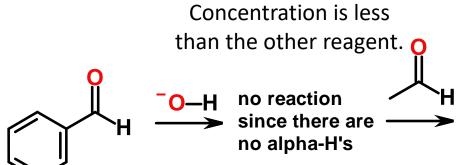
Mixed Aldol Reaction

Added dropwise. Concentration is less Greater than the other reagent. **Concentration** no reaction since there are no alpha-H's Major **Product Minor** Lower **Product Concentration**

Mixed Aldol Reaction

Concentration

 π -system



Added dropwise.

If extended conjugation from C=O to other side is possible, then no heat is

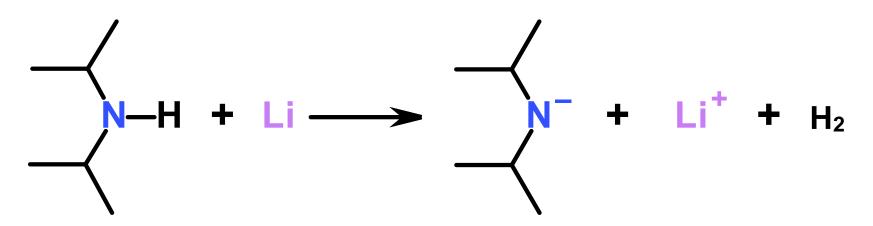
required for elimination!

Greater Major **Concentration** Minor Lower **Product**

Generating LDA

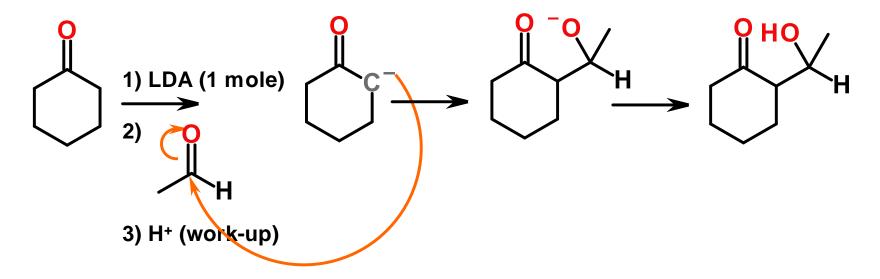
$$H$$
 $O-H + Na \longrightarrow O-Na + H2$

Lithium Di-isopropylamide (LDA)

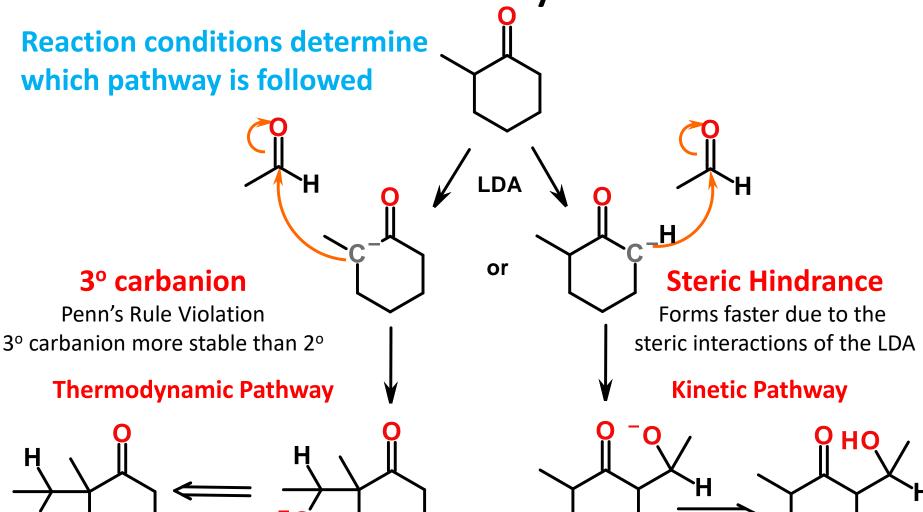


- A. Strong base (N⁻ is not that happy, since N is not that electronegative and doesn't want a negative charge)
- B. Non-nucleophilic, b/c of steric hindrance of isopropyl groups

LDA as a base for H's α to a carbonyl



Kinetic vs Thermodynamic Issues



Examples

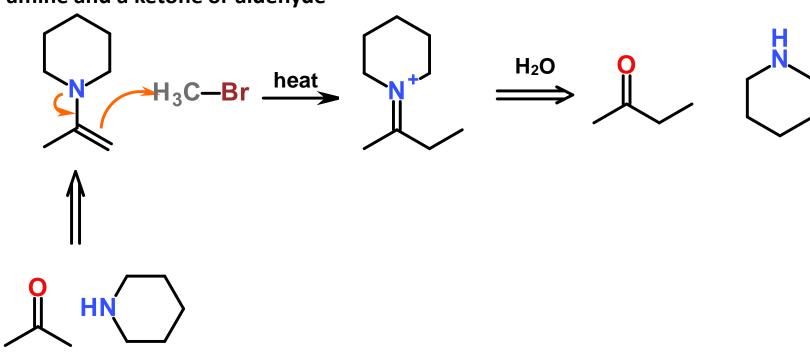
Remember: S_N2 substitution order

Benzylic, allylic > 1° > 2° . 3° , aryl, vinyl do not react

Use Enamines To Get Only A Single Substitution Reaction

Enamine is derived from

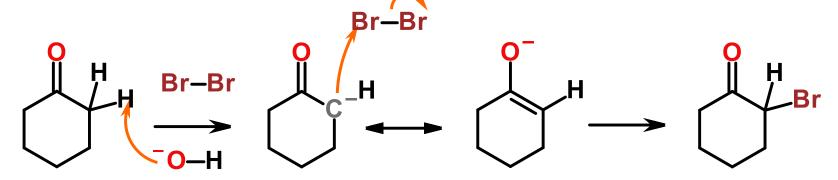
a 2° amine and a ketone or aldehyde



Hell, Vollhardt, Zeliensky Reaction

Reaction of Exchangeable Hydrogens α- to a Carboxylic Acid with Br₂

Bromination α - to a Ketone



Compare these relative reaction rates!

More stable due to electronegativity of Br

Faster Reaction!

Impossible to stop with monohalogenation Brush with basic conditions Brush

Bromination α - to a Ketone

Less stable due to electronegativity of Br

Slower Reaction!

Monohalogenation is possible with acidic conditions

Examples